Oxidative Fragmentations of the 5α - and 5β -Hydroxy-*B*-norcholestan- 3β -yl Acetates

by Mira S. Bjelaković^a), Ljubinka B. Lorenc*^a)^b), Vladimir D. Pavlović^a)^b), Bernard Tinant^c), Jean-Paul Declercq^c), and Jaroslav Kalvoda^d)

^a) Center for Chemistry, ICTM, P.O. Box 473, YU-11001 Belgrade (E-mail: llorenc@helix.chem.bg.ac.yu) ^b) Faculty of Chemistry, University of Belgrade, Studentski trg 12-16, P.O. Box 158, YU-11001 Belgrade

^c) Faculté des Sciences, Département de Chimie, Université catholique de Louvain, Place Pasteur 1,

B-1348 Louvain-la-Neuve

^d) Leimgrubenweg 21, CH-4102 Binningen

In memory of Professor Oskar Jeger, deceased on September 14, 2002

Oxidations of 5α -hydroxy-*B*-norcholestan- 3β -yl acetate (8) with Pb(OAc)₄ under thermal or photolytic conditions or in the presence of iodine afforded only complex mixtures of compounds. However, the HgO/I₂ version of the hypoiodite reaction gave as the primary products the stereoisomeric (*Z*)- and (*E*)-1(10)unsaturated 5,10-seco *B*-nor-derivatives 10 and 11, and the stereoisomeric (*SR*,10*R*)- and (*5S*,10*S*)-acetals 14 and 15 (*Scheme* 4). Further reaction of these compounds under conditions of their formation afforded, in addition, the *A*-nor 1,5-cyclization products 13 and 16 (from 10) and 12 (from 11) (see also *Scheme* 6) and the 6iodo-5,6-secolactones 17 and 19 (from 14 and 15, resp.) and 4-iodo-4,5-secolactone 18 (from 15) (see also *Scheme* 7). Oxidations of 5β -hydroxy-*B*-norcholestan- 3β -yl acetate (9) with both hypoiodite-forming reagents (Pb(OAc)₄/I₂ and HgO/I₂) proceeded similarly to the HgO/I₂ reaction of the corresponding 5α -hydroxy analogue 8. Photolytic Pb(OAc)₄ oxidation of 9 afforded, in addition to the (*Z*)- and (*E*)-5,10-seco 1(10)unsaturated ketones 10 and 11, their isomeric 5,10-seco 10(19)-unsaturated ketone 22, the acetal 5-acetate 21, and 5β ,19-epoxy derivative 23 (*Scheme* 9). Exceptionally, in the thermal Pb(OAc)₄ oxidation of 9, the 5,10-seco ketones 10, 11, and 22 were not formed, the only reaction being the stereoselective formation of the 5,10-seco with the β -oriented epoxy bridge, *i.e.* the (10*R*)-enol ether 20 and (*5S*,10*R*)-acetal 5-acetate 21 (*Scheme* 8). Possible mechanistic interpretations of the above transformations are discussed.

1. Introduction. – Previous studies have shown that alkoxy radicals of type **i** (obtained by oxidation of the 5α - or 5β -hydroxysteroids **1a**,**b** with lead tetraacetate (Pb(OAc)₄) under thermal or photolytic conditions or with hypoiodite-forming reagents) readily undergo β -fragmentation of the C(5)–C(10) bond (*Scheme 1*) to give, *via* the C(10) radical intermediate **ii**, the diastereoisomeric (*Z*)- and (*E*)-1(10)-unsaturated 5-oxo-5,10-secosteroids **2** and **3** in high yield and, depending on the oxidant used, in different proportions [1].

On the other hand, alkoxy radical **iii** generated by similar oxidations of the 14 α -hydroxy-5 α -cholestan-3 β -yl acetate (**4**) reacted preferentially [2] by β -fragmentation of the C(13)-C(14) bond to afford, *via* the C(13) radical **iv**, the methylidene derivative **6** and the (*E*)-12-unsaturated 14-oxo-13,14-seco compound **7** (*Scheme* 2)¹). Exception-

¹) Further reactions of **6** and **7** under conditions of their formation either with oxidant or with some other species present in the reaction solution afforded products of allylic and α -acetoxylation from **6** and a *D*-homo-*C*-nor derivative and a (12*R*,13*R*)-epoxy derivative from **7**.



Scheme 2



ally, the photolytic Pb(OAc)₄ oxidation of **4** resulted, in addition (to *ca.* 20%), in a reversible fragmentation of the alkoxy radical **iii**, involving splitting and recombination of the C(8)–C(14) bond with inversion of the configuration at C(14), followed by formation of the 14 β ,22-epoxy derivative **5**.

In connection with these results, we investigated in the present work oxidative fragmentations of the 5α - and 5β -hydroxy-*B*-norcholestan- 3β -yl acetates **8** and **9** [3][4] which, by analogy with the above examples, could involve as an intermediate the C(10) radical **v** incorporated in a nine-membered ring (*Scheme 3*).

Oxidations of alcohols 8 and 9 were performed with $Pb(OAc)_4$ and hypoioditeforming reagents (hypoiodite reaction [5]) under conditions similar to those previously applied to the compounds 1 and 4.



2. Oxidations of 5α -Hydroxy-*B*-norcholestan- 3β -yl Acetate (8). – 2.1. Attempted Thermal and Photolytic $Pb(OAc)_4$ Oxidation and $Pb(OAc)_4$ Version of the Hypoiodite Reaction of 8. These reactions were performed under typical experimental conditions. However, in all cases they resulted only in complex mixtures of compounds, unresolvable by chromatographic methods.

2.2. HgO/I_2 Version of the Hypoiodite Reaction. The HgO/I₂ oxidation of **8** was carried out with an excess of oxidant in CCl₄ solution by irradiation with a 800-W lamp at 60 V at room temperature for 3 h. The mixture obtained after usual workup was separated by column chromatography (silica gel) to give (*Scheme* 4)²): the stereo-isomeric (*Z*)- and (*E*)-1(10)-unsaturated *B*-nor-5,10-secosteroidal ketones **10** and **11**



²) Yields given in the *Schemes* refer to the isolated pure (by TLC) compounds.

(29.3 and <2%, resp.), the stereoisomeric $(1\alpha,5\beta)$ -A-nor and $(1\beta,5\beta)$ -A-nor cyclization products **12** and **13** (16.7 and 2.5%, resp.), the stereoisomeric (5R,10R)- and (5S,10S)-acetals **14** and **15** (4.0 and 17.2%, resp.) and $(1\beta,5\beta)10\alpha$ -hydroxy cyclization product **16** (6.4%).

The low yield of the (*E*)-seco ketone **11** (as indicated by TLC) was due to its easy cyclization to product **12**. To avoid this transformation, in a similar experiment, the oxidation mixture was saponified with 1% KOH/MeOH at room temperature. This mixture, upon column chromatography (silica gel), afforded the (*Z*)- and (*E*)-3 β -hydroxy derivatives **10a** and **11a** (30.4 and 15.3%, resp.), which were re-acetylated to the corresponding acetates **10** and **11** (Scheme 5)²).



The structures of the oxidation products 10-16 were deduced from their analytical and spectral data (IR, ¹H- and ¹³C-NMR, MS). The proposed configurations of the (*Z*)- and (*E*)-1(10)-unsaturated *B*-nor-seco ketones 10 and 11 were confirmed by X-ray crystallographic determination³).

In the IR spectra of seco ketones **10** and **11** ($C_{28}H_{46}O_3$), the absorption of the original 5*a*-hydroxy group is missing. Instead a new absorption is observed at 1704 and 1678 cm⁻¹, respectively, for the 5-oxo function. The ¹H-NMR spectra⁴) show the following characteristic signals: AcO at 1.68 and 1.80 ppm, respectively, 1 Me–C=C bond at 1.59 and 1.44 ppm, respectively, and 1 olefinic proton at 5.37 and 5.01 ppm, respectively. In the ¹³C-NMR spectra, Me(19) of **10** resonates at 19.3 ppm and that of **11** at 13.6 ppm. The upfield shift of the C(19) resonance of **11**, due to the γ -cis effect (cis-C(2)–C(1)=C(10)–C(19)) [6], exists only in the (*E*)-isomer (conformation **C**, *Fig.* 1), as a result of the shielding interaction of the H_β–C(2) bond with the Me(19) group. This establishes the (*E*)-configuration of **11** and therefore the (*Z*)-configuration of **10**.



Fig. 1. Conformations A and B of 10 and conformation C of 11

³) Results of the crystallographic and conformational analyses (in solution) of **10** and **11** will be presented and discussed in detail in a subsequent publication.

⁴) The ¹H- and ¹³C-NMR spectra of **10** and **11** were recorded in C_6D_6 solution because of instability of the *(E)*-seco ketone **11** in CDCl₃.

In addition, conformations of the nine-membered ring of **10** and **11** in solution were deduced from the ¹H- and ¹³C-NMR spectral analysis³). The results showed that the (*Z*)-stereoisomer **10** and its 3β -OH derivative **10a** exist in solution in two conformations (*Fig. 1*), the major conformer **A** and the minor conformer **B**, while (*E*)-isomer **11** and the corresponding 3β -alcohol **11a** are present in solution in only one conformation (**C**).

Compounds 12, 13, and 16 (for data, see *Exper. Part*) are the secondary products formed by intramolecular cyclization of the (*Z*)- and (*E*)-seco ketones 10 and 11, respectively⁵). Thus, upon stirring in the presence of SiO₂ in toluene solution at room temperature, the (*Z*)-isomer 10 afforded, after 24 h, the cyclization products 13 and 16 in 20.1 and 13.7% yield (besides recovered starting material), while the (*E*)-isomer 11 was transformed, after 1 h, to the cyclization product 12 in almost quantitative yield (*Scheme* 6)²). The configurations of 12 and 13 were deduced from their ¹³C-NMR data.



The C(9) resonance of **13** (41.7 ppm) is shifted upfield with respect to the C(9) resonance of **12** (48.3 ppm) as the result of the γ -gauche effect due to the H_a-C(2) bond. This interaction exists only in the (1 β ,5 β)-isomer, indicating the *cis*-(1 β ,5 β)-configuration for **13**. On the other hand, the difference of the chemical shifts for C(2) of **12** and **13** (31.7 ppm for **12** and 37.0 ppm for **13**) can be attributed to the shielding effect by the HO-C(5) group, which is present only in the (1 α ,5 β)-isomer. This establishes the *trans*-(1 α ,5 β)-configuration for **12**.

Acetal structures for compounds 14 and 15 were determined on the basis of spectral characteristics (see *Exper. Part*). However, the (5R,10R)-configuration for 14 and the (5S,10S)-configuration for 15⁶) were deduced from their ¹³C-NMR data. Also, the

⁵) Intramolecular cyclization of the (*Z*)- and (*E*)-1(10)-unsaturated 5,10-seco ketones **10** and **11** is initiated by protonation of the 5-oxo group and proceeds with participation by the π -electrons of the C(1)=C(10) bond. The mutual neighborhood of the reacting groups in the (*E*)-isomer **11** (*Fig. 1*, conformation **C**) favors this transannular process. For mechanistic and stereochemical aspects of the reaction, see [1][2].

⁶) The other two possible stereoisomers with the (5R,10S)- and the (5S,10R)-configuration, respectively, are too strained to exist as stable compounds.

proposed (5*S*,10*S*)-structure of **15** was unequivocally confirmed by X-ray analysis (see below).

The chemical shift of C(1) of acetal **15** appears at a much higher field (28.9 ppm) than the corresponding chemical shift of acetal **14** (40.9 ppm). In the (5*S*,10*S*)-isomer **15** (*Fig.* 2), the two γ -effects due to the H_{β}-C(8) and H_{β}-C(11) bonds [7] can cause an overall upfield shift of the C(1) signal of 12 ppm relative to the value found for the isomer **14**, where these effects are inexistent.



Fig. 2. Structures of 14 and 15

In the course of the HgO/I₂ oxidation of **8**, it was observed (by TLC) that acetals **14** and **15** were also consumed by reacting with the oxidant. Since the thereby formed compounds could not be separated by chromatographic methods, in an attempt to obtain pure samples of these products for their identification, the acetals **14** and **15** were treated with the HgO/I₂ reagent under conditions similar to those applied to alcohol **8** (*Scheme* 7)²).



Thus, HgO/I₂ reaction of acetals **14** and **15**, respectively, consisted in the oxidative fragmentation of their C(5)-C(6) bond to give the corresponding 6-iodo-5,6-secolactones **17** (from **14**, 58.5%) and **19** (from **15**, 63.7%), while oxidation of acetal **15** involved, in addition, the competing fragmentation of the C(4)-C(5) bond, affording as the minor component the 4-iodo-4,5-secolactone **18** (12.6%). Spectral data

obtained for compounds 17-19 were in complete agreement with the proposed structures. Also, since the oxidative fragmentations of the acetals 14 and 15 took place without change of configuration at the stereogenic centers, the obtained lactones have the (10R)-, (3S,10S)- and (10S)-configurations, respectively.

Differentiation between the 5,6-seco lactones **17** and **19** on one hand and the structurally different 4,5-seco lactone **18** on the other was possible from their ¹H-NMR spectral characteristics.

Appearance of two dd at 3.31 and 3.95 ppm for **17** and at 3.37 and 3.73 ppm for **19** indicates the presence of the iodomethylene (ICH₂(6)) group [8], while in **18**, a m at 3.30 ppm (2 H) suggests the presence of a CH₂ group situated between the I-atom and the AcO function.

3. Oxidations of 5 β -Hydroxy-*B*-norcholestan-3 β -yl Acetate (9). – 3.1. *Thermal* $Pb(OAc)_4$ Oxidation. The thermal $Pb(OAc)_4$ oxidation of 9 was carried out with an excess of oxidant (50%) in the presence of CaCO₃ in boiling benzene for 22 h, when practically all starting material was consumed. Column chromatography (SiO₂) of the resulting mixture afforded only two products (*Scheme* 8)²): the transannular (10*R*)-enol ether **20** (17.0%) and the (5*S*,10*R*)-acetal 5-acetate **21**⁷) (64.9%). Acetal 5-acetate **21** was identified by correlation with the (5*R*,10*R*)-acetal **14**. Saponification of **21** with 1% KOH/MeOH at room temperature and subsequent acetylation of the obtained 3β , 5α -diol with Ac₂O/pyridine yielded **14** of the known structure⁷).

In the IR spectrum of **20**, the 5β -hydroxyl band is missing, instead, new absorptions for an ether bond at 1028 cm⁻¹ and an olefinic C=C bond at 1669 cm⁻¹ are observed. The ¹H-NMR spectrum shows a *s* for Me(19) at 1.25 ppm and a *fd* for 1 olefinic proton at 5.31 ppm. This indicates a transannular 5,10-enol ether structure for compound **20**. The (10*R*)-configuration was deduced from the ¹³C-NMR spectrum: the chemical shift of C(1) at 38.0 ppm is very similar to the value (of *ca.* 40 ppm) observed for the (5*R*,10*R*)-acetal **14** and also the (5*S*,10*R*)-acetal 5-acetate **21**, which possess the β -oriented epoxy bridge⁸).



⁷) Different assignment of the configuration at C(5) of **21** (55) with respect to the configuration of C(5) in acetal **14** (5*R*) is only formally arising from the differencies in *CIP* priority of the groups attached at C(5) in these two compounds.

⁸) The configurational dependence of chemical shifts of C(1) in ¹³C-NMR spectra for stereoisomeric (5R,10R)- and (5S,10S)-acetals is discussed above (see **14** vs. **15**).

3.2. Photolytic $Pb(OAc)_4$ Oxidation. The photolytic $Pb(OAc)_4$ oxidation of **9** was carried out in benzene solution with 3 mol-equiv. of oxidant in the presence of $CaCO_3$ by irradiation with a high-pressure mercury lamp (*Q81*) at room temperature for 5 h. Analysis of products separated by column chromatography (SiO₂) revealed that, under photolytic conditions, the following compounds were formed (*Scheme 9*)²): the (*Z*)-and (*E*)-5,10-seco 1(10)-unsaturated ketones **10** and **11** (10.6 and 16.1%, resp.), their isomeric 5,10-seco 10(19)-unsaturated derivative **22** (17.6%), the (5S,10*R*)-acetal 5-acetate **21** (8.4%), and the 5 β ,19-epoxy derivative **23** (22.1%)⁹).



In the IR spectrum of **22**, the absorption of the 5β -hydroxy group is replaced by an absorption at 1707 cm⁻¹ for the 5-oxo group, while the presence of an exocyclic methylidene group is evident from the absorptions at 3070 and 1639 cm⁻¹. In the ¹H-NMR spectrum, the *s* of Me(19) is replaced by a pair of *s* at 4.93 and 4.96 ppm, and in the ¹³C-NMR spectrum, a *t* appears at 116.9 ppm. The ¹H-NMR spectrum of epoxy derivative **23** shows, instead of a *s* of Me(19), 2 *d* at 3.92 and 4.43 of the CH₂(19) group. The structure is confirmed by the IR absorptions at 1023 and 964 cm⁻¹ for a four-membered ether and by the ¹³C-NMR data (1 *s* at 93.8 ppm for C(5) and 1 *t* at 74.0 ppm for C(19)).

3.3. $Pb(OAc)_4/I_2$ and HgO/I_2 Version of the Hypoiodite Reaction. The Pb(OAc)_4/I_2 reaction [5] of **9** was carried out with a large excess of oxidant in cyclohexane solution by irradiation with a 800-W lamp at 60 V at room temperature for 1 h. Column chromatography (SiO₂) allowed the isolation of the known compounds **10–12**, **14**, and **15** in yields presented in *Scheme 10*²).

The HgO/I₂ version of the hypoiodite reaction of **9** was performed under experimental conditions similar to those applied to the 5 α -hydroxy-*B*-nor derivative **8** (*Scheme 10*). Analysis of the reaction products **10–12**, **14**, and **15** separated by column chromatography (SiO₂) indicated that the HgO/I₂ oxidation of **9** took a similar course as that of the corresponding 5 α -OH analogue **8** (see *Scheme 4*)¹⁰), and also as that of **9** with Pb(OAc)₄/I₂ (see *Scheme 10*).

⁹) Preliminary experiments showed that epoxy derivative **23** is formed from **22** by transannular cycloaddition of the 5-oxo function to the methylene group induced by UV light.

¹⁰) In this reaction with 9, the iodo derivatives 17-19 were also detected by TLC, but not isolated in pure form.



^a) Oxidant Pb(OAc)₄/I₂. ^b) Oxidant HgO/I₂

4. X-Ray Crystal-Structure Analysis of Compound 15. – The structure deduced by spectroscopic means for compound 15 was confirmed by X-ray analysis (*Fig. 3, Table*). H-Atom positions were calculated with the assumption of normal geometry. Lists of fractional atomic coordinates, isotropic thermal parameters, and bond lengths and angles have been deposited at the *Cambridge Crystallographic Data Centre* as supplementary publication No. 188127.

5. Discussion. – The above results indicated that oxidations of the (5α) - and (5β) -*B*- nor alcohols **8** and **9** with the HgO/I₂ reagent and also oxidations of the (5β) -isomer **9**



Fig. 3. X-Ray crystal structure of 15

Table. Crystal Data of Compound 15

Empirical formula	C ₂₈ H ₄₈ O ₄ · ¹ / ₂ CH ₃ OH
M _r	464.68
Temperature	293(2) K
Wavelength	0.71069 Å
Crystal system	monoclinic
Space group	P21/c
Unit cell dimensions	$a = 11.894(4)$ Å, $\alpha = 90^{\circ}$
	$b = 11.414(4)$ Å, $\beta = 95.57^{\circ}$
	$c = 21.059(7)$ Å, $\gamma = 90^{\circ}$
Volume	2845.4(17) Å ³
Ζ	4
Density (calc.)	1.085 Mg/m^3
Absorption coefficient	0.071 mm^{-1}
F (000)	1028
Crystal size	0.22 imes 0.16 imes 0.14 mm
θ range for data collection	$1.89 - 24.74^{\circ}$
Index ranges	$-13 \le h \le 13, -13 \le k \le 13, -24 \le l \le 24$
Reflections collected	20050
Independent reflections	8559 (R(int) = 0.067)
Refinement method	full-matrix least-squares on Fsqd
Data, restraints, parameters	8559, 47, 615
Goodness-of-fit on F^2	1.046
Final R indices $(I > 2\sigma(I))$	$R_1 = 0.0628, wR_2 = 0.1762$
R indices (all data)	$R_1 = 0.0769, wR_2 = 0.1900$
Absolute structure parameter	-0.3(12)
Extinction coefficient	0.049 (4)
Largest diff. peak and hole	0.348 and $-0.223 \text{ e} \cdot \text{\AA}^{-3}$

by the Pb(OAc)₄/ $h\nu$ and Pb(OAc)₄/ I_2 methods proceed, as expected, *via* the C(10)centered alkyl radical **F**. It is generated by homolysis of the O-X bond (X = I or Pb(OAc)₃) in the primarily formed species **D**, followed by β -fragmentation of the C(5)-C(10) bond in the alkoxy radical **E** thus obtained (*Scheme 11*). However, the fate of the nine-membered-ring C(10) radical **F** is considerably different from that of



2130

the corresponding ten-membered-ring analogue **ii** (results shown in *Scheme 1*) and also from the nine-membered-ring C(13) radical **iv** (see *Scheme 2*). In the present case, stabilization of the C(10) radical **F** proceeds by two independent reaction pathways: *a*) it takes place by elimination of H_{α} – or H_{β} –C(1) to produce the stereoisomeric (*Z*)and (*E*)-seco ketones **10** and **11**, or by elimination of a H-atom from the Me(19) group to give the 10(19)-unsaturated seco ketone **22**; or *b*) it involves a non-stereoselective attack of the O-atom of the 5-oxo group at C(10) to give finally the (*5R*,10*R*)-acetal **14** (the minor product with the β -oriented epoxy bridge) and the (*5S*,10*S*)-isomer **15** (the major product with the α -oriented epoxy bridge).

However, the thermal Pb(OAc)₄ oxidation of 5β -hydroxy-*B*-nor compound **9** (see *Scheme 8*) deviates from the above mechanistic scheme: 1) in this reaction, the 5,10seco ketones **10**, **11**, or **22** are not found among oxidation products, and 2) the reaction results exclusively in the stereoselective formation of the 5,10-ethers with the β oriented epoxy bridge¹¹), *i.e.*, the (10*R*)-enol 5,10-ether **20** and the (5*S*,10*R*)-acetal 5acetate **21**¹²). This indicates that the C(10) radical **F** does not participate in the thermal Pb(OAc)₄ oxidation of compound **9**¹³) and suggests that not only radicals but also various cationic species should be considered as possible intermediates in the above reaction¹⁴).

The authors from Belgrade acknowledge the financial support by the *Ministry of Science, Technology, and Development of Serbia* (part of the project 'Synthesis and Chemical Transformations of Steroidal and Modified Steroidal Molecules', project code 1702).

Experimental Part

1. General. Column chromatography (CC): silica gel 0.04-0.063 mm; FC = flash chromatography. TLC: monitoring and separation of products with silica gel *G* (*Stahl*), detection with aq. 50% H₂SO₄ soln. M.p.: uncorrected. IR Spectra: *Perkin-Elmer 337* spectrophotometer; $\tilde{\nu}$ in cm⁻¹. NMR Spectra: *Varian Gemini-200* (¹H at 200, ¹³C at 50 MHz); CDCl₃ or C₆D₆ soln. at r.t.; SiMe₄ as internal standard; δ in ppm, *J* in Hz; *f*: 'fine structure' of signals. Mass spectra: *Finnigan MAT 8230*; ionization energy 70 eV; *m/z* (rel. intensity in %).

2. Oxidation of **8**: Hypoiodite HgO Oxidation of $(5\alpha, 3\beta)$ -B-Nor-5-cholestane-3,5-diol 3-Acetate (**8**). A stirred suspension of **8** (500 mg, 1.157 mmol), yellow HgO (1.595 g, 7.36 mmol), and I₂ (2.13 g, 8.39 mmol) in anh. CCl₄ (50 ml) was irradiated with a 800-W Osram-Halogen-Bellaphot lamp (at 60 V) at r.t. for 3 h. The mixture was filtered, the filtrate was washed successively with aq. Na₂S₂O₃ soln., aq. NaHCO₃ soln., and H₂O, dried (Na₂SO₄), and evaporated, and the resulting mixture (605 mg) was separated by FC (SiO₂ (80 g), toluene/AcOEt 97:3 (\rightarrow **10**, **11**), 95:5 (\rightarrow **12**, **13**), 70:30 (\rightarrow **14**, **15**), 50:50 (\rightarrow **16**)): **10** (146 mg, 29.3%), **10/11** (10 mg, 2%), **12** (83 mg, 16.7%), **13** (12.4 mg, 2.5%), **14** (20.7 mg, 4.0%), **15** (89.2 mg, 17.2%), and **16** (33.2 mg, 6.4%).

¹¹) As mentioned above, oxidations of 5-hydroxy-*B*-nor compounds **8** and **9** with hypoiodite-forming reagents (Pb(OAc)₄/I₂ and/or HgO/I₂), which proceed by the intermediacy of the C(10) radical **F**, give the (5*R*,10*R*)-acetal **14** with the β -oriented epoxy bridge as the minor product.

¹²) Rosenthal et al. [9][10] obtained similar results upon oxidation of 6β -methyl-*B*-nor- 5β -androstane- 3β , 5β ,17 β -triol 3,17-diacetate with Pb(OAc)₄ under thermal conditions. With this substrate, the corresponding enol 5,10-ether and the 5,10-acetal 5-acetate with β -epoxy bridge were also the only isolated reaction products.

¹³) Rosenthal et al. considered as possible mechanisms of the thermal Pb(OAc)₄ oxidation of the androstane B-nor-5β-hydroxysteroid a direct rearrangement of the alkoxy radical of type E to the corresponding epoxy derivatives by 1,2-migration of the 10-alkyl group [10].

¹⁴) This topic will be discussed in detail in a subsequent publication.

 $\begin{bmatrix} 1(10)Z_3\beta_j - 3-(Acetyloxy) - B-nor-5, 10-secocholest - 1(10)-en-5-one (10)^{15} \end{bmatrix}: M.p. 81-82^{\circ} \text{ (from MeOH)}. \\ \begin{bmatrix} \alpha \end{bmatrix}_{D}^{20} = +11.7 \ (c=1.2, \text{ CHCl}_3). \text{ IR (KBr): 1740, 1704, 1236, 1029. }^{1}\text{H-NMR} \ (C_6D_6): 0.61 \ (s, Me(18)); 0.94 \\ (d, Me(26), Me(27)); 0.96 \ (d, Me(21)); 1.59 \ (s, Me(19)); 1.68 \ (s, AcO); 2.20 \ (dd, J=4.7, 13, H_{\beta}-C(4)); 2.57 \\ (t, J=3.3, H_{\beta}-C(2)); 5.37 \ (fdd, J\approx 1, 6, 11, H-C(1)); 5.65 \ (m, H-C(3)). \\ ^{13}\text{C-NMR} \ (C_6D_6): 209.6 \ (s, C(5)); \\ 169.3 \ (s, MeCOO); 143.3 \ (s, C(10)); 121.5 \ (d, C(1)); 72.7 \ (d, C(3)); 56.7 \ (d, C(17)); 53.2 \ (d, C(14)); 48.5 \\ (t, C(4)); 44.2 \ (t, C(6)); 43.7 \ (d, C(9)); 43.0 \ (s, C(13)); 39.9 \ (t, C(24)); 39.5 \ (t, C(12)); 39.4 \ (d, C(8)); 36.5 \\ (t, C(22)); 36.1 \ (d, C(20)); 30.3 \ (t, C(2)); 28.3 \ (d, C(25)); 28.1 \ (t, C(16)); 26.5 \ (t, C(11)); 25.2 \ (t, C(15)); 24.3 \\ (t, C(23)); 23.0 \ (q, C(27)); 22.7 \ (q, C(26)); 20.7 \ (q, MeCOO); 19.3 \ (q, C(19)); 18.9 \ (q, C(21)); 11.8 \ (q, C(18)). \\ \text{CI-MS: 431 (23, [M+1]^+), 413 (100, [M-18+1]^+), 371 (46, [M-60+1]^+), 353 (70, [M-18-60+1]^+). \\ \text{Anal. calc. for } C_{78}H_{46}O_3 \ (430.34): C 78.09, H 10.77; found: C 77.80, H 10.51. \\ \end{bmatrix}$

Separation of (Z)- and (E)-Seco Ketones **10** and **11**. A stirred suspension of **8** (300 mg, 0.694 mmol), yellow HgO (957 mg, 4.42 mmol), and I₂ (1.278 g, 5.03 mmol) in anh. CCl_4 (25 ml) was irradiated with two 800-W Osram-Halogen-Bellaphot lamps (at 60 V) at r.t. for 1 h. The resulting mixture (370 mg), isolated after the usual workup, was dissolved in MeOH (50 ml) and hydrolyzed with 1% KOH/MeOH (10 ml). The resulting mixture was left at r.t. for 1 h, evaporated at r.t. to *ca*. 10 ml, diluted with H₂O, and extracted with Et₂O. The org. layer was washed with H₂O until neutral, dried (Na₂SO₄), and evaporated. The residue (310 mg) was subjected to FC (SiO₂ (6 g), toluene/AcOEt 96 :4): **11a** (41.3 mg, 15.3%) and **10a** (82 mg, 30.4%).

 $[1(10)Z_3\beta]$ -3-Hydroxy-B-nor-5,10-secocholest-1(10)-en-5-one (10a): ¹H-NMR (C₆D₆; two nine-membered-ring conformations I/II 74:26): 0.63 (I), 0.73 (II) (parts of 2s, Me(18)); 1.64 (I), 1.69 (II) (parts of 2s, Me(19)); 4.39 (I), 4.14 (II) (parts of 2m, H-C(3)); 5.57 (I), 5.05 (II) (dd, J (I) = 5.3, 10.9, J(II) = 6.1, 11.5, H-C(1)).

[1(10)E,3 β]-3-Hydroxy-B-nor-5,10-secocholest-1(10)-en-5-one (**11a**): ¹H-NMR (C₆D₆): 0.63 (*s*, Me(18)); 0.94 (*d*, Me(26), Me(27)); 0.97 (*d*, Me(21)); 1.53 (*s*, Me(19)); 2.68 (*dd*, J = 10.3, 14.3, H_{β}-C(2)); 4.29 (*fq*, J = 7, H-C(3)); 5.20 (*m*, H-C(1), OH). ¹³C-NMR (C₆D₆): 212.0 (*s*, C(5)); 144.8 (*s*, C(10); 127.5 (*d*, C(1)); 76.0 (*d*, C(3)); 56.9 (*d*, C(9)); 56.7 (*d*, C(17)); 54.6 (*d*, C(14)); 48.8 (*t*, C(4)); 45.2 (*t*, C(6)); 43.3 (*s*, C(13)); 42.5 (*d*, C(8)); 39.9 (*t*, C(24)); 39.6 (*t*, C(12)); 36.5 (*t*, C(22)); 36.0 (*d*, C(20)); 35.0 (*t*, C(2)); 28.3 (*t*, C(16)); 28.2 (*d*, C(25)); 25.7 (*t*, C(15)); 25.0 (*t*, C(11)); 24.2 (*t*, C(23)); 22.9 (*q*, C(27)); 22.7 (*q*, C(26)); 18.9 (*q*, C(21)); 12.9 (*q*, C(19)); 12.1 (*q*, C(18)).

The (Z)-ketone **10a** (82 mg) was acetylated with Ac₂O/pyridine 1:1 (3.5 ml) for 3 h. Purification by FC (SiO₂, 1 g) gave **10** (88.1 mg, 97.0%), which was recrystallized from MeOH (yield: 68.2 mg, 75.1%).

The (*E*)-ketone **11a** (41.3 mg) was acetylated with Ac₂O/pyridine 1 : 1 (2 ml) for 8 h : **11** (40 mg, 87.4%). The crude material was purified by FC (SiO₂ (0.5 g), toluene/AcOEt 95 : 5) and recrystallized from MeOH (yield: 16 mg, 35.0%).

 $\begin{bmatrix} I(10)-\text{E},3\beta J\text{-}3\text{-}Acetoxy\text{-}B\text{-}nor\text{-}5,10\text{-}secocholest\text{-}I(10)\text{-}en\text{-}5\text{-}one} (\mathbf{11})\text{:} \text{M.p. 116}-118^{\circ} (from MeOH). } [a]_{D}^{2D} = \\ + 33.8 \ (c = 1, \text{ CHCl}_3)\text{. IR (KBr): } 1725, 1678, 1638, 1248, 1033. } ^{1}\text{H}\text{-}\text{NMR (C}_6\text{D}_6)\text{:} 0.67 \ (s, Me(18))\text{;} 0.94 \\ (d, Me(26), Me(27))\text{;} 0.97 \ (d, Me(21))\text{;} 1.44 \ (s, Me(19))\text{;} 1.80 \ (s, AcO)\text{;} 2.18 \ (td, J = 5.6, 16.2, H_a - C(2))\text{;} 2.44 \\ (d, J = 4.8, H_a - C(4))\text{;} 2.57 \ (dd, J = 7.8, 13.6, H_\beta - C(4))\text{;} 5.01 \ (dd, J = 4.7, 10.9, \text{H} - C(1))\text{;} 5.15 \ (fquint. J = 5.7, 1.1, \text{H} - C(3))\text{.} ^{13}\text{C}\text{-}\text{NMR (C}_6\text{D}_6\text{):} 202.9 \ (s, C(5))\text{;} 169.9 \ (s, MeCOO)\text{;} 145.1 \ (s, C(10))\text{;} 123.5 \ (d, C(1))\text{;} 74.4 \ (d, C(3))\text{;} 57.4 \ (d, C(9))\text{;} 56.9 \ (d, C(17))\text{;} 54.7 \ (d, C(14))\text{;} 47.2 \ (t, C(4))\text{;} 46.8 \ (t, C(6))\text{;} 43.5 \ (s, C(13))\text{;} 41.9 \ (d, C(8))\text{;} 39.9 \\ (t, C(24))\text{;} 39.7 \ (t, C(12))\text{;} 36.5 \ (t, C(22))\text{;} 36.1 \ (d, C(20))\text{;} 31.8 \ (t, C(2))\text{;} 28.3 \ (t, C(16))\text{;} 28.2 \ (d, C(25))\text{;} 25.7 \\ (t, C(15))\text{;} 25.1 \ (t, C(11))\text{;} 24.3 \ (t, C(23))\text{;} 22.9 \ (q, C(27))\text{;} 22.7 \ (q, C(26))\text{;} 20.9 \ (q, MeCOO)\text{;} 18.9 \ (q, C(21))\text{;} 13.6 \ (q, C(19))\text{;} 12.2 \ (q, C(18))\text{.} \text{CI-MS: } 431 \ (12, [M + 1]^+)\text{.} \text{Anal. calc. for } C_{28}H_{46}O_3 \ (430.34)\text{:} C 78.09, \text{H 10.77}\text{;} found: C 78.42, \text{H 11.04}. \end{aligned}$

 $(1a, 3\beta, 5\beta)$ -5($10 \rightarrow 1$)*Abeo*-B-*norcholest-10*(19)-*ene-3,5-diol* 3-*Acetate* (12): Oil. $[a]_{20}^{20} = +11.5$ (c = 2.2, CHCl₃). IR (CHCl₃): 3493, 3090, 3072, 1738, 1650, 1246, 892. ¹H-NMR (CDCl₃): 0.70 (s, Me(18)); 0.86 (d, Me(26), Me(27)); 0.92 (d, Me(21)); 2.06 (s, AcO); 4.71, 4.92 (2s, CH₂(19)); 5.24 (m, H–C(3)). ¹³C-NMR (CDCl₃): 171.0 (s, MeCOO); 149.4 (s, C(10)); 105.8 (t, C(19)); 79.0 (s, C(5)); 73.4 (d, C(3)); 56.1 (d, C(17)); 56.0 (d, C(14)); 53.2 (d, C(1)); 48.3 (d, C(9)); 45.3 (t, C(4)); 43.1 (s, C(13)); 41.0 (t, C(6)); 39.4 (t, C(24)); 39.4 (t, C(12)); 39.4 (d, C(8)); 36.0 (t, C(22)); 35.6 (d, (20)); 31.7 (t, C(2)); 28.1 (t, C(16)); 27.9 (d, C(25)); 24.4 (t, C(15)); 23.8 (t, C(23)); 23.7 (t, C(11)); 22.7 (q, C(27)); 22.4 (q, C(26)); 21.2 (q, MeCOO); 18.5 (q, C(21); 12.1 (q, C(18)). CI-MS: 431 (6, [M + 1]⁺), 353 (100, [M - 18 - 60 + 1]⁺).

¹⁵ (Z)-Seco ketone 10 exists in solution in two conformations A and B (see Fig. 1); A/B ca. 85:15. The given ¹H- and ¹³C-NMR data refer to the main conformation.

 $(1\beta_3\beta_5\beta_9)-5(10 \rightarrow 1)Abeo$ -B-norcholest-10(19)-ene-3,5-diol 3-Acetate (13): Oil. $[a]_D^{20} = -1.2$ (c = 2.0, CHCl₃). IR (CHCl₃): 3452, 3083, 1738, 1641, 1247, 893. ¹H-NMR (CDCl₃): 0.68 (s, Me(18)); 0.86 (d, Me(26), Me(27)); 0.92 (d, Me(21)); 2.04 (s, AcO); 2.84 (dd, J = 9, 12, H–C(1)); 4.80, 4.88 (2s, CH₂(19)); 5.07 (fq, $J \approx 7$, H–C(3)). ¹³C-NMR (CDCl₃): 170.8 (s, MeCOO); 149.1 (s, C(10)); 109.3 (t, C(19)); 79.6 (s, C(5)); 72.7 (d, C(3)); 56.1 (d, C(17)); 55.9 (d, C(14)); 54.0 (d, C(1)); 47.5 (s, C(13)); 43.0 (t, C(4)); 41.7 (d, C(9)); 40.2 (t, C(6)); 39.4 (t, C(12)); 39.3 (t, C(24)); 38.2 (d, C(8)); 37.0 (t, C(2)); 36.1 (t, C(22)); 35.7 (d, C(20)); 28.2 (t, C(16)); 27.9 (d, C(25)); 24.4 (t, C(15)); 23.9 (t, C(11)); 23.8 (t, C(23)); 22.8 (q, C(27)); 22.5 (q, C(26)); 21.2 (q, MeCOO); 18.6 (q, C(21)); 12.0 (q, C(18)). CI-MS: 431 (2, $[M+1]^+$), 353 (100, $[M-18-60+1]^+$).

 $(3\beta,5R,10R)$ -5,10-Epoxy-B-nor-5,10-secocholestane-3,5-diol 3-Acetate (14): Oil. $[a]_D^{20} = +1.1$ (c = 1.7, CHCl₃). IR (CHCl₃): 3425, 1735, 1367, 1246, 757. ¹H-NMR (CDCl₃): 0.70 (s, Me(18)); 0.87 (d, Me(26), Me(27)); 0.90 (d, Me(21)); 1.15 (s, Me(19)); 2.07 (s, AcO); 5.01 (m, H–C(3)). ¹³C-NMR (CDCl₃): 170.3 (s, MeCOO); 97.0 (s, C(5)); 76.0 (s, C(10)); 72.3 (d, C(3)); 56.3 (d, C(17)); 56.0 (d, C(14)); 48.7 (d, C(9)); 47.4 (t, C(4)); 42.7 (s, C(13)); 40.9 (t, C(11)); 39.7 (t, C(12)); 39.4 (t, C(24)); 39.3 (t, C(6)); 36.1 (t, C(22)); 35.7 (d, C(20)); 31.6 (d, C(8)); 28.8 (t, C(2)); 28.1 (t, C(16)); 27.9 (d, C(25)); 25.8 (q, C(19)); 23.8 (3t, C(11), C(15), C(23)); 22.8 (q, C(27)); 22.5 (q, C(26)); 21.4 (q, MeCOO), 18.6 (q, C(21)); 12.1 (q, C(18)). MS: 448 (7, M^+), 388 (40, [M - 60]⁺), 135 (100%).

(3β,58,108)-5,10-Epoxy-B-nor-5,10-secocholestane-3,5-diol 3-Acetate (**15**): M.p. 112–113° (from acetone/MeOH). $[a]_D^{20} = +29.5$ (c = 1.5, CHCl₃). IR (KBr): 3422, 1734, 1367, 1250, 757. ¹H-NMR (CDCl₃): 0.72 (s, Me(18)); 0.86 (d, Me(26), Me(27)); 0.90 (d, Me(21)); 1.15 (s, Me(19)); 2.01 (s, AcO); 2.39 (m, CH₂(6)); 3.26 (br. s, OH); 5.29 (m, H–C(3)). ¹³C-NMR (CDCl₃): 170.4 (s, MeCOO), 95.8 (s, C(5)); 77.1 (s, C(10)); 69.1 (d, C(3)); 56.2 (d, C(17)); 54.9 (d, C(14)); 51.3 (d, C(9)); 42.8 (t, C(4)); 42.3 (s, C(13)); 39.6 (t, C(12)); 39.4 (t, C(24)); 39.4 (t, C(6)); 36.0 (t, C(22)); 35.6 (d, C(20)); 32.8 (d, C(8)); 28.9 (t, C(11)); 28.1 (t, C(16)); 27.9 (d, C(25)); 27.9 (t, C(2)); 26.9 (q, C(19)); 23.7 (t, C(23)); 23.6 (t, C(15)); 22.7 (q, C(27)); 22.6 (t, C(11)); 22.5 (q, C(26)); 21.3 (q, MeCOO), 18.6 (q, C(21)); 12.1 (q, C(18)). MS: 448 (4, M⁺), 388 (59, [M - 60]⁺), 371 (10, [M - 60 - 17]⁺). Anal. calc. for C₂₈H₄₈O₄ (448.36): C 74.95, H 10.78; found: C 75.01, H 11.01.

 $(1\beta_3\beta_5\beta_7,10R)$ - $5(10 \rightarrow 1)$ Abeo-B-norcholestane-3,5,10-triol 3-Acetate (**16**): Oil. $[a]_{20}^{20} = +23.7$ (c = 1.2, CHCl₃). IR (CHCl₃): 3441, 1718, 1260. ¹H-NMR (CDCl₃): 0.69 (s, Me(18)); 0.86 (d, Me(26), Me(27)); 0.91 (d, Me(21)); 1.38 (s, Me(19)); 2.02 (s, AcO); 5.0 (dq, J = 2, 7, H-C(3)). ¹³C-NMR (CDCl₃): 170.8 (s, MeCOO); 78.8 (s, C(5)); 73.1 (s, C(10)); 71.3 (d, C(3)); 56.0 (d, C(17)); 55.8 (d, C(14)); 55.1 (d, C(1)); 49.6 (t, C(4)); 46.4 (d, C(9)); 42.6 (s, C(13)); 40.3 (t, C(6)); 39.4 (t, C(24)); 39.4 (t, C(12)); 36.0 (t, C(22)); 35.7 (d, C(20)); 33.8 (d, C(8)); 32.9 (t, C(2)); 28.1 (t, C(16)); 27.9 (d, C(25)); 24.9 (q, C(19)); 24.1 (t, C(15)); 23.8 (t, C(23)); 22.7 (q, C(27)); 22.5 (q, C(26)); 21.2 (q, MeCOO); 20.5 (t, C(11)); 18.6 (q, C(21)); 12.0 (q, C(18)). CI-MS: 431 ($[M-18+1]^+$), 371 (100, $[M-60-18+1]^+$).

Treatment of **10** with SiO_2 in *Toluene*. A mixture of **10** (100 mg) and SiO_2 (3 g) in toluene (10 ml) was stirred for 24 h at r.t. Workup gave a mixture (95 mg), which was separated by CC (SiO_2 (10 g), toluene/AcOEt 97:3 \rightarrow 70:30): starting material **10** (52 mg, 52.0%), **13** (20.1 mg, 20.1%), and **16** (14.3 mg, 13.7%).

Treatment of **11** with SiO_2 in Toluene. A mixture of **11** (10 mg) and SiO_2 (300 mg) in toluene (1 ml) was stirred for 30 min at r.t. Silica gel was removed by filtration and washed several times with toluene. The filtrates were combined and evaporated. The residue was chromatographed (SiO₂ (1 g), toluene/AcOEt 95:5): **12** (9.9 mg, 98.8%).

Hypoiodite HgO Oxidation of (5R,10R)-*Acetal* **14**. A stirred suspension of **14** (100 mg, 0.22 mmol), yellow HgO (319 mg, 1.47 mmol), and I₂ (425 mg, 1.67 mmol) in CCl₄ (60 ml) was irradiated with a 800-W *Osram-Halogen-Bellaphot* lamp (at 60 V) at r.t. for 1 h. The resulting mixture (115 mg), isolated after the usual workup, was separated by FC (SiO₂ (20 g), toluene/AcOEt 85:15): **17** (75 mg, 58.5%), and starting material **14** (10 mg, 10%).

(4S,7R)-4-(Acetyloxy)-7-[(1R,3aS,4S,5S,7aR)-1-[(1R)-1,5-dimethylhexyl]octahydro-4-(iodomethyl)-7a $methyl-1H-inden-5-yl]-7-methyloxepan-2-one (17): Oil. <math>[a]_{D}^{20} = +11.9 \ (c = 0.7, CHCl_3)$. IR (neat): 1737, 1235, 1147, 1033. ¹H-NMR (CDCl_3)¹⁶): 0.75 (s, Me(18)); 0.87 (d, Me(26), Me(27)); 0.93 (d, Me(21)); 1.39 (s, Me(19)); 2.07 (s, AcO); 3.24 (m, H_a-C(4)); 3.31 (dd, J = 2.4, 9.6, H-C(6)); 3.95 (dd, J = 2.0, 9.8, H-C(6)); 5.18 (m, H-C(3)). ¹³C-NMR (CDCl_3)¹⁶): 170.2 (s, C(5)); 169.9 (s, MeCOO); 86.8 (s, C(10)); 66.3 (d, C(3)); 56.0 (d, C(17)); 54.3 (d, C(14)); 43.2 (d, C(9)); 43.0 (t, C(4)); 41.8 (s, C(13)); 39.4 (t, C(24)); 39.1 (t, C(12)); 36.7 (d, C(8)); 35.9 (t, C(22)); 35.6 (d, C(20)); 33.8 (t, C(1)); 28.0 (t, C(16)); 28.0 (d, C(25)); 27.5 (t, C(2)); 24.8 (q, C(19)); 24.7 (t, C(15)); 23.7 (t, C(23)); 23.4 (t, C(11)); 22.8 (q, C(27)); 22.5 (q, C(26)); 21.1 (q, MeCOO);

16) Steroid numbering.

20.1 (*t*, C(6)); 18.5 (*q*, C(21)); 12.8 (*q*, C(18)). CI-MS: 575 (100, $[M+1]^+$), 515 (68, $[M-60+1]^+$), 447 ($[M-128+1]^+$), 387 ($[M-60-128+1]^+$).

Hypoiodite HgO Oxidation of (5S,10S)-*Acetal* **15**. As described for **17**, with **15** (180 mg, 0.40 mmol), HgO (574 mg, 2.65 mmol), I_2 (765 mg, 3.01 mmol), and CCl₄ (60 ml) (for 2 h). The residue (206 mg) was separated by FC (SiO₂ (30 g), toluene/AcOEt 90:10): **18** (29 mg, 12.6%) and **19** (147 mg, 63.7%).

(4\$,4a\$,6aR,7R,9a\$,9b\$)-4-[(3\$)-3-(Acetyloxy)-4-iodobutyl]-7-[(1R)-1,5-dimethylhexyl]-4,4a,5,6,6a,7,8,9,-9a,9b-decahydro-4,6a-dimethylcyclopenta[f][2]benzopyran-2(1H)-one (**18**): Oil. $[a]_{D}^{20} = -9.6 (c = 1.0, CHCl_3)$. IR (neat): 1737, 1239, 1024. ¹H-NMR (CDCl₃)¹⁶): 0.70 (s, Me(18)); 0.89 (d, Me(26), Me(27)); 0.95 (d, Me(21)); 1.39 (s, Me(19)); 2.10 (s, AcO); 2.60 (dd, J = 4.9, 17.3, H_β-C(6)); 3.30 (m, CH₂(4)); 4.72 (quint., J = 5.4, H-C(3)). ¹³C-NMR (CDCl₃)¹⁶): 170.8 (s, C(5)); 170.4 (s, MeCOO); 86.3 (s, C(10)); 72.5 (d, C(3)); 55.9 (d, C(17)); 55.8 (d, C(14)); 49.3 (d, C(9)); 42.5 (s, C(13)); 39.4 (t, C(24)); 39.1 (t, C(12)); 36.0 (t, C(22)); 35.6 (t, C(6)); 35.6 (d, C(20)); 31.6 (t, C(1)); 30.8 (d, C(8)); 28.1 (t, C(2)); 27.9 (t, C(16)); 27.9 (d, C(25)); 26.2 (q, C(19)); 23.7 (t, C(23)); 23.4 (t, C(15)); 22.8 (t, C(11)); 22.7 (q, C(27)); 22.5 (q, C(26)); 21.0 (q, MeCOO); 18.6 (q, C(21)); 12.5 (q, C(18)); 7.5 (t, C(4)). CI-MS: 575 (100, [M + 1]⁺).

(4\$,7\$)-4-(Acetyloxy)-7-[(1R,3a\$,4\$,5\$,5,7aR)-1-[(1R)-1,5-dimethylhexyl]octahydro-4-(iodomethyl)-7amethyl- IH-inden-5-yl]-7-methyloxepan-2-one (**19** $): Oil. <math>[a]_{D}^{20} = +20.5 \ (c = 1.6, CHCl_3)$. IR (neat): 1750, 1720, 1246, 1153, 1022. ¹H-NMR (CDCl_3)¹⁶): 0.72 (*s*, Me(18)): 0.87 (*d*, Me(26), Me(27)); 0.90 (*d*, Me(21)); 1.46 (*s*, Me(19)); 2.07 (*s*, AcO); 2.88 (*dd*, *J* = 2.1, 15.3, H_β-C(4)); 3.13 (*fdd*, *J* = 5.3, 15.3, H_a-C(4)); 3.37 (*dd*, *J* = 2.4, 9.8, H-C(6)); 3.73 (*dd*, *J* = 2.2, 9.8, H-C(6)); 5.17 (*m*, H-C(3)). ¹³C-NMR (CDCl_3)¹⁶): 169.9 (*s*, C(5)); 169.3 (*s*, MeCOO); 86.1 (*s*, C(10)); 66.4 (*d*, C(3)); 55.9 (*d*, C(17)); 54.3 (*d*, C(14)); 49.6 (*d*, C(9)); 42.0 (*t*, C(4)); 41.7 (*s*, C(13)); 39.4 (*t*, C(24)); 39.0 (*t*, C(12)); 36.3 (*d*, C(8)); 35.9 (*t*, C(22)); 35.6 (*d*, C(20)); 27.9 (*d*, C(25)); 27.9 (*t*, C(16)); 27.6 (2*t*, C(1), C(2)); 24.0 (*q*, C(19)); 23.7 (2*t*, C(15), C(23)); 23.7 (*t*, C(11)); 22.8 (*q*, C(27)); 22.5 (*q*, C(26)); 21.1 (*q*, MeCOO); 18.5 (*q*, C(21)); 17.9 (*t*, C(6)); 12.7 (*q*, C(18)). CI-MS: 575 (67, [*M*+1]⁺), 515 (100, [*M*-60+1]⁺), 447 ([*M*-128+1]⁺), 387 ([*M*-60-128+1]⁺).

3. Oxidations of **9**. 3.1. Thermal $Pb(OAc)_4$ Oxidation of $(3\beta,5\beta)$ -B-Norcholestane-3,5-diol 3-Acetate (**9**). A suspension of **9** (200 mg, 0.463 mmol), Pb(OAc)_4 (307.8 mg, 0.694 mmol), and CaCO₃ (69.5 mg, 0.694 mmol) in dry benzene (25 ml) was heated under reflux with stirring for 22 h (monitoring by the starch/KI test). The mixture was diluted with Et₂O and filtered through a *Celite* pad, and the insoluble precipitate was thoroughly washed with Et₂O. The combined filtrate was washed with H₂O, sat. aq. NaHCO₃ soln. and H₂O, dried (Na₂SO₄), and evaporated and the residue (240 mg) separated by FC (SiO₂ (25 g), toluene/AcOEt 95:5): **20** (34 mg, 17.0%) and **21** (148 mg, 64.9%).

 $(3\beta, 10R)$ -5,10-Epoxy-B-nor-5,10-secocholest-4-en-3-ol Acetate (**20**): Oil. $[a]_{20}^{20}$ + 32.8 (c = 1.4, CHCl₃). IR (neat): 1737, 1669, 1242, 1028. ¹H-NMR (CDCl₃): 0.72 (s, Me(18)); 0.87 (d, Me(26), Me(27)); 0.91 (d, Me(21)); 1.25 (s, Me(19)); 2.06 (s, AcO); 2.61 (dd, J=6.9, 13.5, H_a-C(6)); 4.86 (fq, J=6, H-C(3)); 5.31 (fd, J=2.5, H-C(4)). ¹³C-NMR (CDCl₃): 170.7 (s, MeCOO); 151.2 (s, C(5)); 118.2 (d, C(4)); 81.0 (s, C(10)); 73.4 (d, C(3)); 56.1 (d, C(17)); 55.9 (d, C(14)); 53.4 (d, C(9)); 42.7 (s, C(13)); 39.8 (t, C(6)); 39.6 (t, C(12)); 39.4 (t, C(24)); 38.0 (t, C(11)); 36.1 (t, C(22)); 36.0 (d, C(8)); 35.7 (d, C(20)); 28.1 (t, C(16)); 28.0 (d, C(25)); 27.6 (t, C(2)); 24.2 (t, C(15)); 24.1 (t, C(11)); 23.8 (t, C(23)); 23.6 (q, C(19)); 22.8 (q, C(27)); 22.5 (q, C(26)); 21.3 (q, MeCOO); 18.7 (q, C(21)); 118. (q, C(18)). CI-MS: 431 (73, $[M+1]^+$), 371 (100, $[M-60+1]^+$).

(3β,5S,10R)-5,10-Epoxy-B-nor-5,10-secocholestane-3,5-diol Diacetate (21): Oil. $[a]_{D}^{20} = -28.9$ (c = 1.7, CHCl₃). IR (neat): 1737, 1240, 1053, 1015. ¹H-NMR (CDCl₃): 0.69 (s, Me(18)); 0.87 (d, Me(26), Me(27)); 0.90 (d, Me(21)); 1.22 (s, Me(19)); 2.02 (s, AcO); 2.05 (s, AcO); 2.39 (dd, $J = 2.8, 14.8, H_{\beta} - C(6)$); 2.63 (dd, $J = 8.2, 14.8, H_{\alpha} - C(6)$); 4.98 (m, H-C(3)). ¹³C-NMR (CDCl₃): 170.5 (s, MeCOO-C(3)); 168.9 (s, MeCOO-C(5)); 105.4 (s, C(5)); 78.0 (s, C(10)); 70.8 (d, C(3)); 56.2 (d, C(17)); 55.8 (d, C(14)); 46.8 (d, C(9)); 44.9 (t, C(4)); 42.6 (s, C(13)); 39.9 (t, C(1)); 39.4 (2t, C(12), C(24)); 37.8 (t, C(6)); 36.0 (t, C(22)); 35.6 (d, C(20)); 31.1 (d, C(8)); 29.3 (t, C(2)); 28.0 (t, C(16)); 27.9 (d, C(25)); 25.8 (q, C(19)); 23.7 (2t, C(15), C(23)); 23.5 (t, C(11)); 22.7 (q, C(27)); 22.4 (q, C(26)); 22.4 (q, MeCOO-C(5)); 21.3 (q, MeCOO-C(3)); 18.5 (q, C(21)); 11.9 (q, C(18)). CI-MS: 491 (3, [M+1]⁺), 431 (14, [M-60+1]⁺), 371 (100, [M-2×60+1]⁺).

3.2. Photolytic $Pb(OAc)_4$ Oxidation of **9**. To a soln. of **9** (400 mg, 0.926 mmol) in anh. benzene (80 ml), Pb(OAc)_4 (1.23 g, 2.775 mmol) and CaCO₃ (0.278 g) were added. The vigorously stirred mixture was irradiated with a high-pressure Hg lamp *Q81 Hanau* at r.t. for 5 h (monitoring by the starch/KI test). The mixture was worked up as above, leaving a mixture (495 mg), which was separated by FC (SiO₂ (90 g), toluene/AcOEt 95:5): **10** (42 mg, 10.6%), **11** (64 mg, 16.1%), **22** (70 mg, 17.6%), **21** (38 mg, 8.4%), and **23** (88 mg, 22.1%).

 (3β) -3-(Acetyloxy)-B-nor-5,10-secocholest-10(19)-en-5-one (22): Oil. $[a]_D^{20} = +34.8 \ (c = 1.1, CHCl_3)$. IR (neat): 3070, 1739, 1707, 1639, 1241, 1025. ¹H-NMR (CDCl_3): 0.79 (s, Me(18)); 0.87 (d, Me(26), Me(27)); 0.91 (d, Me(21)); 2.03 (s, AcO); 2.99 (dd, J = 10.3, 13.5, H_{\beta}-C(4)); 4.93, 4.96 (2s, CH₂(19)); 5.20 (m, H-C(3)).

¹³C-NMR (CDCl₃): 210.3 (*s*, C(5)); 170.0 (*s*, MeCOO); 148.8 (*s*, C(10)); 116.9 (*t*, C(19)); 71.9 (*d*, C(3)); 56.3 (*d*, C(17)); 54.7 (*d*, C(14)); 53.8 (*d*, C(9)); 48.9 (*t*, C(4)); 47.0 (*t*, C(6)); 42.9 (*s*, C(13)); 39.5 (*t*, C(12)); 39.4 (*t*, C(24)); 36.5 (*d*, C(8)); 36.0 (*t*, C(22)); 35.7 (*d*, C(20)); 33.3 (*t*, C(2)); 29.0 (*t*, C(1)); 28.6 (*t*, C(16)); 27.9 (*d*, C(25)); 27.7 (*t*, C(11)); 24.8 (*t*, C(15)); 23.7 (*t*, C(23)); 22.7 (*q*, C(27)); 22.5 (*q*, C(26)); 21.3 (*q*, MeCOO); 18.6 (*q*, C(21)); 12.1 (*q*, C(18)). CI-MS: 431 (100, $[M+1]^+$), 371 (50, $[M-60+1]^+$).

 $(3\beta,5\beta)$ -5,19-*Epoxy*-B-*norcholestan-3-ol Acetate* (23): Oil. $[a]_D^{20} = +11.9$ (c = 1.7, CHCl₃). IR (neat): 1739, 1242, 1023, 964. ¹H-NMR (CDCl₃): 0.65 (s, Me(18)); 0.87 (d, Me(26), Me(27)); 0.94 (d, Me(21)); 2.01 (s, AcO); 3.92, 4.43 (2d, J = 6, CH₂(19)); 5.23 (m, H–C(3)). ¹³C-NMR (CDCl₃): 170.1 (s, MeCOO); 93.8 (s, C(5)); 74.0 (t, C(19)); 68.3 (d, C(3)); 56.9 (d, C(17)); 55.6 (d, C(14)); 55.4 (d, C(9)); 46.9 (s, C(10)); 46.7 (t, C(4)); 43.5 (s, C(13)); 42.3 (d, C(8)); 39.5 (t, C(12)); 39.4 (t, C(24)); 39.3 (t, C(6)); 36.1 (t, C(22)); 35.6 (d, C(20)); 28.4 (t, C(16)); 27.9 (d, C(25)); 25.2 (t, C(1)); 24.8 (t, C(15)); 23.7 (t, C(23)); 22.7 (q, C(27)); 22.5 (q, C(26)); 21.2 (t, C(11)); 21.1 (q, MeCOO); 20.1 (t, C(2)); 18.7 (q, C(21)); 12.0 (q, C(18)). CI-MS: 431 (14, [M +1]⁺), 371 (100, [M – 60 + 1]⁺).

Saponification of Diacetate **21**. Diacetate **21** (50 mg) was dissolved in MeOH (5 ml) and hydrolyzed with 1% KOH/MeOH (0.5 ml) at r.t. for 1 h. The mixture was evaporated, the residue diluted with H_2O and extracted with Et_2O , and the org. layer was washed with H_2O , dried (Na_2SO_4) and evaporated; 42 mg of diol. The crude diol was acetylated with Ac_2O (0.2 ml) in anh. pyridine (0.2 ml) overnight at r.t. Usual workup gave a residue, which was subjected to FC (SiO_2 (2 g), toluene/AcOEt 95:5): **14** (41 mg, 82%). Spectral data: identical to those of **14** (see above).

3.3. *Hypoiodite Pb(OAc)*₄ *Oxidation of* **9**. A mixture of **9** (200 mg, 0.46 mmol), Pb(OAc)₄ (841 mg, 1.897 mmol), and I₂ (670 mg, 2.640 mmol) in cyclohexane (80 ml) was stirred and irradiated with a 800-W *Osram-Halogen-Bellaphot* lamp (at 60 V) at r.t. for 1 h. The mixture was filtered off, and the filtrate was washed successively with aq. NaHCO₃ soln. and H₂O, dried (Na₂SO₄), and evaporated. The residue (250 mg) was separated by FC (SiO₂ (40 g), toluene/AcOEt 96:4 (\rightarrow **10–12**), 80:20 (\rightarrow **14**, **15**)): **10** (43 mg, 21.6%), **11** (42 mg, 21.1%), **12** (5 mg, 2.5%), **14** (18 mg, 8.7%), and **15** (52 mg, 25.1%).

3.4. Hypoiodite HgO Oxidation of **9**. A stirred suspension of **9** (360 mg, 1.83 mmol), yellow HgO (1.148 g, 5.30 mmol), and I₂ (1.530 g, 6.03 mmol) in anh. CCl₄ (30 ml) was irradiated with a 800-W Osram-Halogen-Bellaphot lamp (at 60 V) at r.t. for 1 h. The mixture was filtered, the filtrate was washed successively with aq. Na₂S₂O₃ soln., aq. NaHCO₃ soln., and H₂O, dried (Na₂SO₄), and evaporated, and the resulting mixture (410 mg), separated by FC (SiO₂ (60 g), toluene/AcOEt 97:3 (\rightarrow **10–12**), 80:20 (\rightarrow **14**, **15**)): **10** (68 mg, 19.1%), **11** (54.7 mg, 15.3%), **12** (9.3 mg, 2.6%), **14** (20 mg, 5.4%), and **15** (54.8 mg, 14.7%).

REFERENCES

- M. Lj. Mihailović, Lj. Lorenc, M. Gašić, M. Rogić, A. Melera, M. Stefanović, *Tetrahedron* 1966, 22, 2345;
 M. Akhtar, S. March, *J. Chem. Soc.* (C) 1966, 937; M. Lj. Mihailović, Lj. Lorenc, J. Foršek, V. Pavlović, M. Dabović, J. Kalvoda, *J. Serb. Chem. Soc.* 1989, 54, 645.
- [2] M. S. Bjelaković, Lj. B. Lorenc, V. D. Pavlović, M. Lj. Mihailović, B. Tinant, J. P. Declercq, J. Kalvoda, *Helv. Chim. Acta* 1999, 82, 707.
- [3] J. Joska, J. Fajkoš, F. Šorm, Collect. Czech. Chem. Commun. 1963, 28, 82; M. S. Bjelaković, V. D. Pavlović, Lj. B. Lorenc, J. Serb. Chem. Soc. 2002, 67, 69.
- [4] J. Joska, J. Fajkoš, Collect. Czech. Chem. Commun. 1963, 28, 621.
- [5] J. Kalvoda, K. Heusler, Synthesis 1971, 525.
- [6] H. Fuhrer, Lj. Lorenc, V. Pavlović, G. Rihs, G. Rist, J. Kalvoda, M. Lj. Mihailović, Helv. Chim. Acta 1981, 64, 703.
- [7] N. K. Wilson, J. B. Stothers, in 'Topics in Stereochemistry', Vol. 8, Eds. E. L. Eliel and N. L. Allinger, Wiley-Interscience, New York, London, 1974, p. 25, and ref. cit. therein.
- [8] H. Suginome, S. Yamada, J. Org. Chem 1985, 50, 2489.
- [9] D. Rosenthal, C. F. Lefler, M. E. Wall, *Tetrahedron Lett.* 1965, 3203.
- [10] D. Rosenthal, C. F. Lefler, M. E. Wall, Tetrahedron 1967, 23, 3583.

Received November 15, 2002